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Drivers for the Chromatographic Separation of Enantiomers

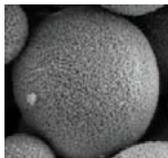
- Retention is strongly dependent on thermodynamics of partition of analytes between mobile and stationary phase
- Separation is driven by the specific interactions of analytes with the chiral selector
- To enhance the separation most **non-specific interactions with the support need to be minimized**
- In reality, the type of support dictates:
 - residual interactions between solutes and stationary phase (e.g. silanol interactions)
 - the chemistry that can be used to bond the selectors
 - the operating range of the CSP
- Carbon has the broadest operating range of existing supports (in reversed phase: pH 1-14, T up to 200 °C)
- Objective: Develop CSPs based on Carbon**

The Base ZirChrom-Carb Particle



Particles are coated with a thin layer of carbon using a chemical vapor deposition process

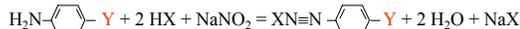
Characteristic	Property
Surface Area (m ² /g)	22
Pore Volume (cc/g)	0.13
Pore Diameter (Å)	250-300
Porosity	0.45
Density (g/cc)	5.8 (2.5x silica)
Particle Diameter (µ)	5.0



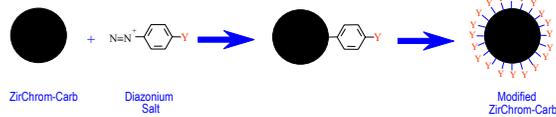
1µm 25000X

Surface Modification of Carbonaceous Particles

Step 1



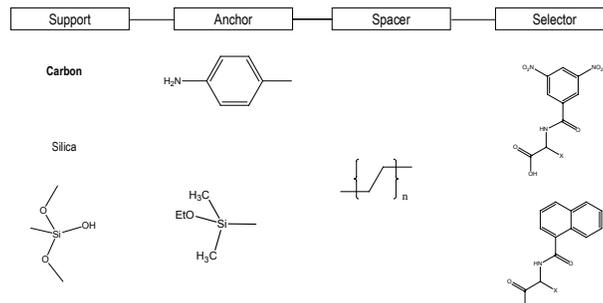
Step 2



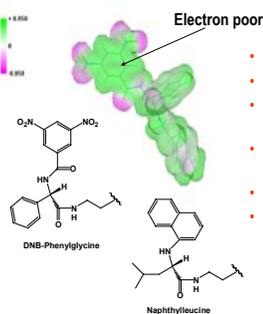
Current Applications:

- DIAMOND BOND-C18** HPLC Columns (ZirChrom Separations Inc)
- Cabot Inkjet Colorants

Building a Chiral Stationary Phase

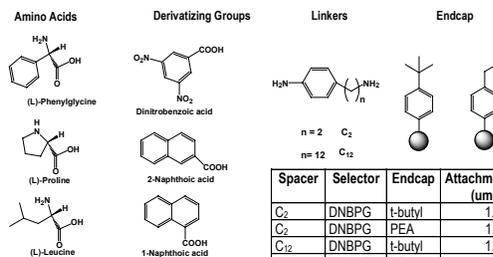


Pirkle-Type Chiral Selectors



- Concepts invented by Prof. William Pirkle in the early 80's
- Products mainly offered by Regis Technologies
- Small organic molecules such as derivatized amino acids are used as chiral selectors
- Separation mechanism is based on p-p interactions and hydrogen-bonding between target molecules and CSPs
- p-electron acceptor:** 3,5-Dinitrobenzoyl (DNB)
- p-electron donor:** Naphthyl

Chemistries That Are Being Explored



Spacer	Selector	Endcap	Attachment Level (µmol/m ²)
C ₂	DNBPG	t-butyl	1.75
C ₂	DNBPG	PEA	1.82
C ₁₂	DNBPG	t-butyl	1.89
C ₁₂	DNBPG	PEA	1.59
C ₂	DNBPro	t-butyl	N/A
C ₂	DNBPro	PEA	N/A
C ₁₂	DNBPro	t-butyl	1.15
C ₁₂	DNBPro	PEA	1.85
C ₂	DNBPG	None	2.29
C ₂	DNBPro	None	2.82

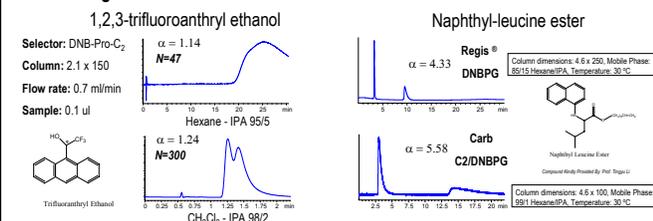
- A range of derivatized amino acids were successfully attached covalently to ZirChrom-Carb particles
- The surface density of attached chiral selectors compares well with reported values for silica-based CSPs with similar selectors

Chiral Chromatography Results

1. Separation Factors (α)

Analyte	C ₂ -DNBPG	C ₂ -DNBPro	Regis® DNBPG
Trifluoroanthyrol ethanol	1.18	1.30	1.51
1,1'-bis-2-naphthol	1.20	1.23	1.82
1-naphthyl-leucine ester	5.58	1.08	4.33

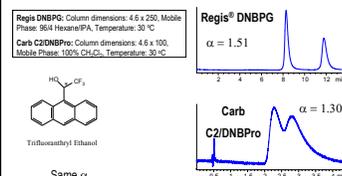
2. Chromatograms



Mobile phase selection is very important for achieving the best separation. Methylene chloride gave better results than hexane on the carbon CSPs for some analytes.

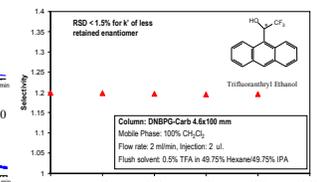
For some analytes the carbon based CSPs gave better separations than the silica-based columns with the same functionality, indicating that the choice of support can affect the separation.

3. Efficiency of the Carbon Chiral Columns



The efficiency of these first-of-a-kind carbon-based CSPs is poorer than that of comparable silica-based CSPs, which adversely affects the resolution of enantiomers of some compounds.

4. Stability of the Carbon Based Columns



Carbon-based columns are ultra-stable, even when exposed to 0.5% TFA, because of the ultra-stable covalent bonding to the surface.

Conclusions

- Novel chiral stationary phases based on bonded carbon clad zirconia chromatographic particles have been developed
- The effect of spacer length and precap type is not as important as the surface density and type of the chiral selector**
- For some molecules the carbon based columns have higher selectivity than similar silica-based commercial columns
- The carbon based columns are stable under more aggressive HPLC conditions, promising more flexibility in chiral method development
- FUTURE WORK:**
 - The eluent composition that works best with carbon based supports needs to be optimized
 - The efficiency of the carbon-based chiral columns must be improved

Acknowledgement: NIH SBIR Grant 1 R43 HL70334-01